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## Synthesis of New Chiral Catalysts, Substituted 2-Phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-quinolines, for the Enantioselective Addition of Diethylzinc to Benzaldehyde

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**Abstract**: New chiral ligands, substituted 2-phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-quinolines, were prepared and the catalyzed reaction of diethylzinc with benzaldehyde was studied under various conditions to give 1-phenyl-1-propanol with up to 91% ee.

Pyridine derivatives designed as ligands for metal complexes with predetermined helical chirality have recently been synthesized. 1,2 The development of this synthetic method opens up the possibility to prepare a series of ligands that are useful in enantioselective catalysis. Many 2-(1-hydroxyalkyl)pyridines have proved to be efficient catalysts in the asymmetric addition of dialkylzinc to aldehydes. We report here the preparation of new chiral ligands, 2-phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-8-quinolinone 4, 5 and 2-phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-8-quinolinone 6 according to scheme 1. The three ligands have been used in the addition reaction of diethylzinc to benzaldehyde.

Scheme 1: a)  $^{1}$ O<sub>2</sub>, Ac<sub>2</sub>O, pyridine, 4-dimethylaminopyridine, meso-tetraphenylporphyrine, 20°C, 16h, 95%, Lit.  $^{4}$  b) phenacylpyridinium bromide<sup>5</sup>, NH<sub>4</sub>OAc, HOAc, 100°C, 20h, 60%, Lit.  $^{6}$  c) i) HOOAc, HOAc, 70°C, 24 h; ii) isobutyric anhydride, 100°C, 8h, 65%, Lit.  $^{7}$  d) NaOH, methanol-water, reflux, 1h, 95%, Lit.  $^{7}$  c) chromic acid, acetone-water, 20°C, 7h, 80%, Lit.  $^{7}$ 

The pinocarvone 1 was obtained by  ${}^{1}O_{2}$  oxidation of (-)-α-pinene in a quantitative yield. 1 reacts with phenacylpyridinium bromide 5 to give 2-phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-quinoline 2 in 60% yield. 6 By oxidation of 2 to N-oxide with peracetic acid followed by Boekelheide-rearrangement with isobutyric anhydride we obtained (2-Phenyl-5,6,7,8-tetrahydro-6,6-dimethyl-5,7-methano-8-quinolinyl) isobutyrate 3 as diastereomeric mixture of 3' and 3" in a 65 / 35 ratio. 7 3' could be isolated by recrystallisation in hexane and 3" by column chromatography (silica gel, hexane-ethyl acetate (10:1)). The ligands 4 and 5 were

obtained by hydrolysis of 3' or 3'', respectively, and the ligand 6 by oxidation of 4 and / or 5 with chromic acid. The ee values of the ligands 4 and 5 were determined by H NMR of esters derived from (1S)-camphanoyl chloride.

Having prepared these ligands, we studied their properties as catalysts for the enantioselective addition of diethylzinc to benzaldehyde (Scheme 2 and Table 1). Ligand 5 proved to be the best catalyst in the series (entry 3). Ligand 4, that differs from 5 only in the configuration on C8, induces the opposite enantiomer of 1-phenyl-1-propanol (S-enantiomer), but has a lower catalytic activity especially with regard to enantioselectivity (entry 1). At 24°C the reaction is faster than at 0°C but the temperature has no significant effect on the enantiomeric purity (entry 2, 3). The use of a polar solvent is detrimental to the efficiency of the catalyst (entry 5).

Table 1: Enantioselective addition of diethylzing to benzaldehyde

Entry <sup>a</sup>	Ligand	Reac. temp. (°C)	Reac. time (h)	Yield (%)b	ee (%)b	Config.c
1	4	0	24	75	28	S
2	5	0	9	78	91	R
3	5	24	3	98	91	R
4	6	24	21	96	75	R
5d	6	24	24	20e	< 1	-

a) All reactions were carried out in toluene-hexane (1:1) if not otherwise indicated; b) Determined by GC analysis using a  $\beta$ -DEXTM110 capillary column from the firm SUPELCO; c) Determined by comparison with the chromatograms of enantiomerically pure samples of R- and S-1-phenyl-1-propanol; d) THF instead of toluene was used; e) The reaction gave also 22% benzyl alcohol.

Representative procedure (entry 3): A dried 10-ml Schlenk flask was charged with 13.3 mg (0.05 mmol) of 5, 2 ml of dry toluene and 1 mmol (106 mg) of benzaldehyde. To this stirred solution 2 ml of a 1 M solution of diethylzinc in hexane (2 mmol) was added dropwise over a 10-min period. After 3 h at 20°C the reaction was quenched with 5 ml of 1 M HCl. The reaction mixture was extracted with dichloromethane and the organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent the remaining oil was analyzed by GC to determine the conversion ratio of the benzaldehyde and the ee of the 1-phenyl-1-propanol. The full characterization of the ligands and an extended study of their catalytic activity are in progress.

## References

- 1. Hayoz, P.; von Zelewsky, A.; Tetrahedron Lett., 1992, 33, 5165.
- 2. Hayoz, P.; von Zelewsky, A.; J. Am. Chem. Soc., 1993, 115, 5111.
- 3. (a) Bolm, C.; Zehnder, M.; Bur, D.; Angew. Chem., 1990, 102, 206; (b) Bolm, C.; Schlingloff, G.; Harms, K.; Chem. Ber., 1992, 125, 1191; (c) Chelucci, G.; Soccolini, F.; Tetrahedron: Asymmetry, 1992, 3, 1235; (d) Ishizaki, M.; Fujita, K.; Shimamoto, M.; Hoshino, O.; Tetrahedron: Asymmetry, 1994, 5, 411; (e) Ishizaki, M.; Hoshino, O.; Tetrahedron: Asymmetry, 1994, 5, 1901; (f) Macedo, E.; Moberg, C.; Tetrahedron: Asymmetry, 1995, 6, 549; (g) Nakano, H.; Kumagai, N.; Kabuto, C.; Matsuzaki, H.; Hongo, H.; Tetrahedron: Asymmetry, 1995, 6, 1233.
- 4. Mihelich, E. D.; Eickhoff, D. J.; J. Org. Chem., 1983, 48, 4135.
- 5. Bamberger, E.; Ber. Deutsch. Chem. Ges., 1887, 20, 3344.
- 6. Kröhnke, F.; Synthesis, 1976, 1.
- 7. Reimann, E.; Reitz, R.; Liebigs Ann. Chem., 1975, 1081.